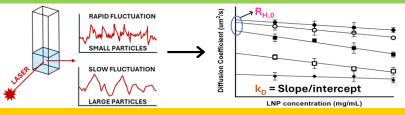
### Behavior of Lipid Nanoparticles in Aqueous Formulations: Measure of LNP Attraction and Repulsion

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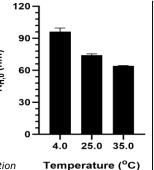
# **What Does Dynamic Light Scattering Measure?**

### Hydrodynamic Radius (R<sub>H</sub>)

$$R_{H,0} = \frac{kT}{6 \pi \eta D_0}$$

Hydrodynamic radius at infinite dilution  $(R_{H,0})$  excludes interparticle interactions.

k = Boltzmann's constant, T = Absolute temperature,  $\eta = Bulk \ viscosity, D_0 = Diffusion coefficient at infinite dilution$ 

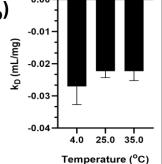


Attraction and Repulsion (k<sub>D</sub>)

$$D = D_0 (1 + k_D c)$$

k<sub>D</sub> values: **positive** (Repulsion); **negative** (Attraction)

 $D = Diffusion coefficient, D_0 = Diffusion coefficient at$ infinite dilution, c = concentration

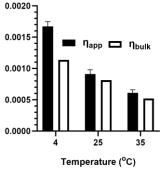


## Apparent Viscosity ( $\eta_{app}$ )

$$\eta_{app} = \frac{kT}{6 \pi R_{H,0}} D_0$$

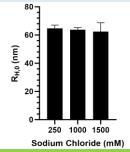
Viscosity experienced by LNP at LNP-water interface ( $\eta_{app}$ ) differs from  $\eta_{\text{bulk}}$ .

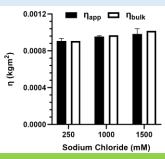
k = Boltzmann's constant, T = Absolute temperature, $R_{H.0}$  = Hydrodynamic radius at infinite dilution

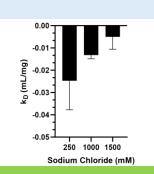


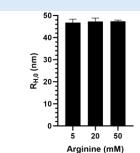
Apparent Viscosity (η<sub>app</sub>)

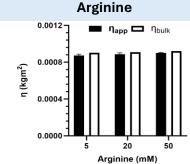


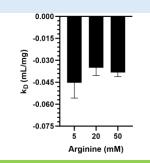


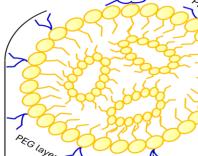












#### References

 Sodium chloride and Arginine have no measurable effects on Lipid nanoparticles (LNP) behavior in solution.

**Conclusions** 

- The apparent viscosity  $(\eta_{app})$  decreases with increasing temperature.
- When a change in the  $R_{H,0}$  of LNPs is unlikely, the apparent viscosity  $(\eta_{ann})$  could change.
- Wright et al., Anal. Chem., 2025 (doi.org/10.1021/acs.analchem.4c06089).
- Francis et al., Int.J.Pharm, 2025 (Under Review).
- Francis et al., 2025 (Manuscript in progress).

Hydrodynamic Radius (RH)